



VTT

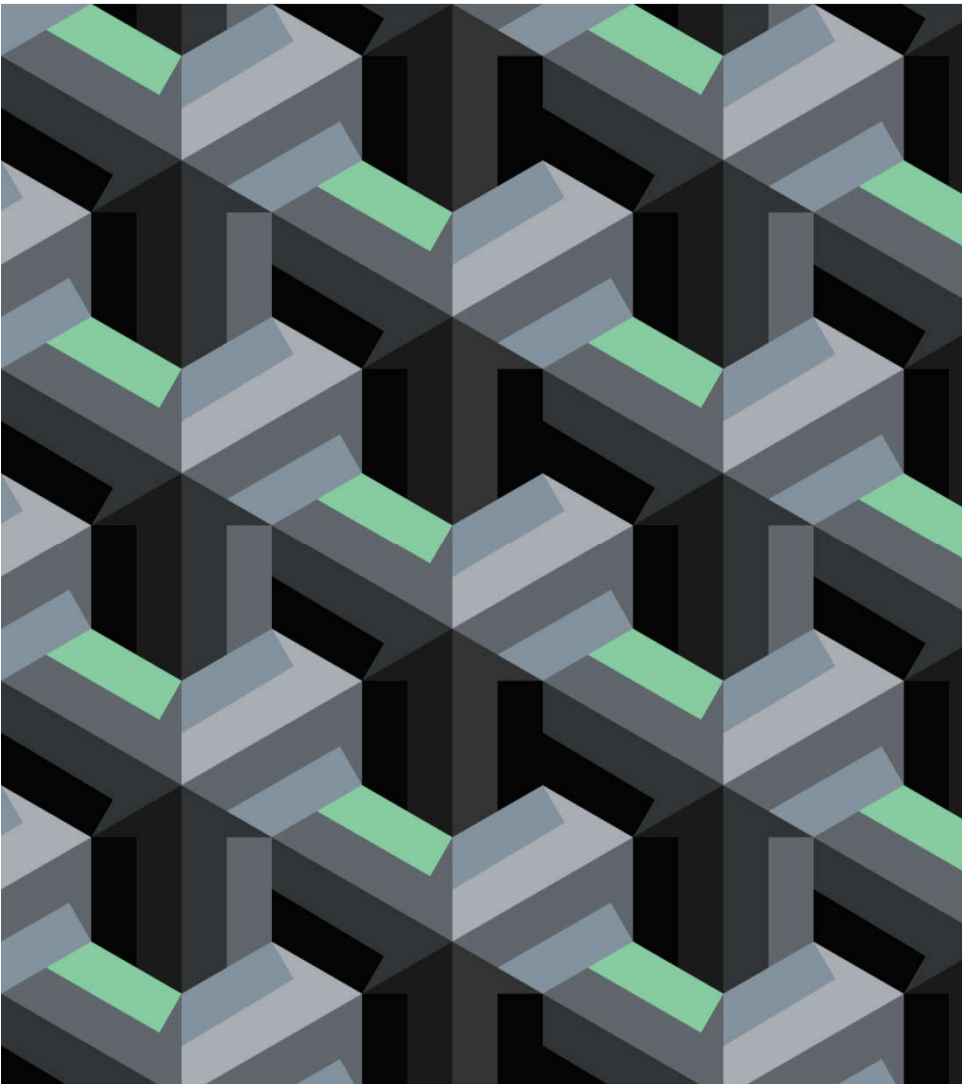
High-Temperature Experimental Techniques for Nuclear Fuel Separate Effect Tests

30/10/2019

SYP - Janne Heikinheimo

Outline

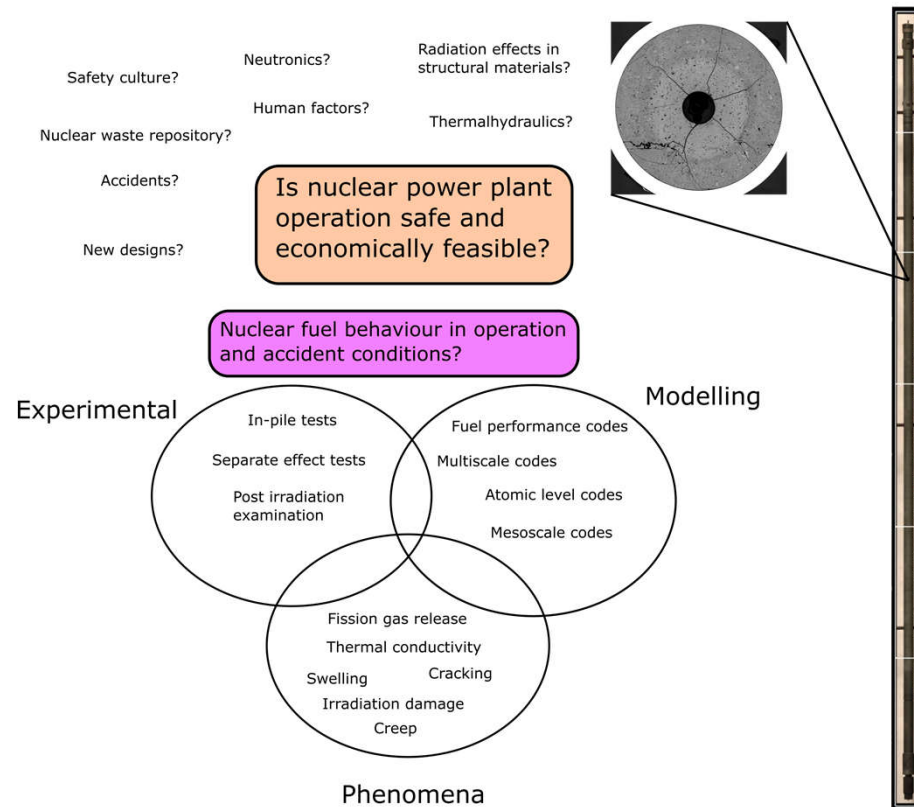
- Why separate effect tests are required for nuclear fuels?
- Furnace for UO₂ separate effect tests
- Uranium oxide properties – oxygen partial pressure
- Example: nuclear fuel aerosol particle preparation
- Future plans

The left side of the slide features a complex, abstract geometric pattern. It is composed of interlocking, three-dimensional-looking shapes in shades of grey, blue, and green, set against a black background. The pattern has a crystalline or lattice-like appearance, with some elements resembling interconnected cubes or prisms.

Why separate effect tests are required for nuclear fuels?

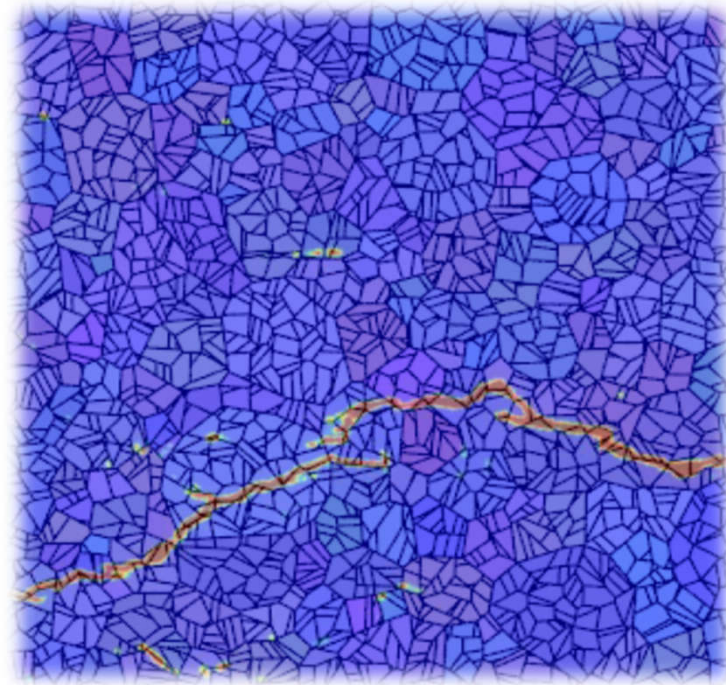
Full-scale experiments and macroscopic modelling

- Full-scale experiments in reactor conditions
 - The behaviour corresponds the actual fuel behaviour
 - The origin of many phenomena cannot be explicitly studied
- Macroscopic modelling
 - Works well with the validated cases
 - Cannot model accurately phenomena arising from the microstructure
 - May not be reliable in predicting the fuel behaviour beyond the validation data



Separate effect tests and microstructural modelling

- Separate effect tests connected to microstructural characterization
 - Recognize the cause for a phenomenon
 - Improve model predictions
- Microstructural and/or multiscale modelling
 - Takes into account microstructural development and inhomogeneities in the structure
 - Utilize more physics based approach to fuel simulations compared to the macroscopic modelling
 - Computationally heavy



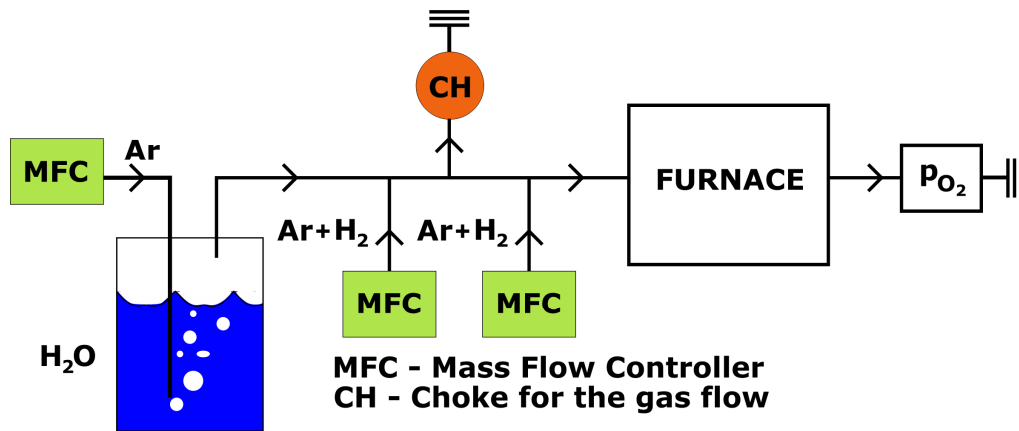
Experimental

Furnace for UO₂ separate effect tests at VTT Centre for Nuclear Safety

- Sample preparation:
 - Sintering at 1700 °C in H₂ + Ar atmosphere to obtain stoichiometric UO₂
- Possible separate effect tests for UO₂ pellets:
 - Oxygen partial pressure adjustment
 - Temperature effects: thermal diffusion of elements, precipitation, cracking
- Microscopy characterization of the samples with SEM or TEM



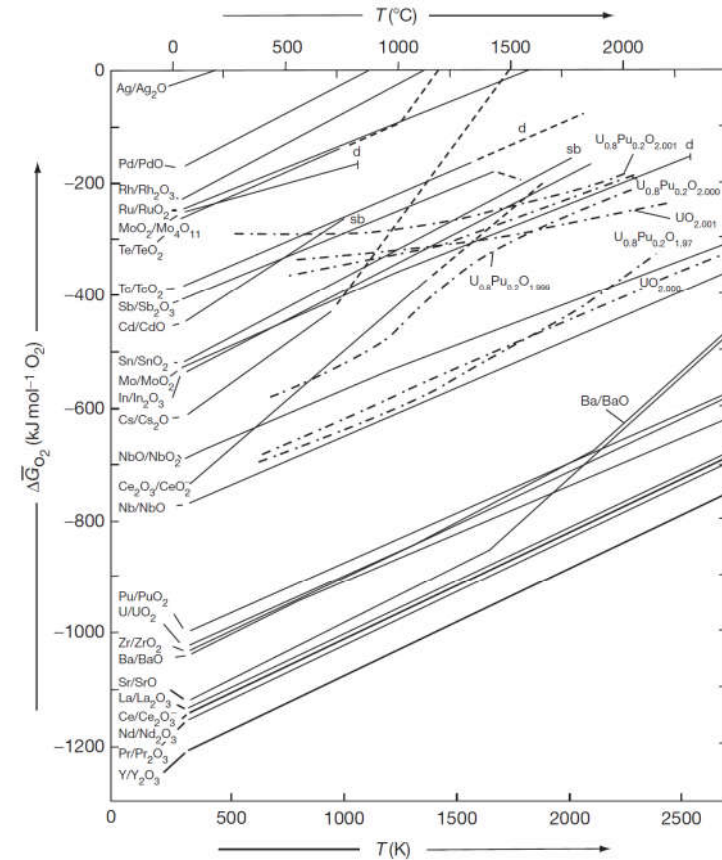
Uranium oxide properties – oxygen partial pressure



$$p_{O_2} = \left(\frac{p_{H_2O}}{p_{H_2}} \exp\left(-\frac{\Delta G_0}{RT}\right) \right)^2$$

$$p_{O_2} = 6e-9 \text{ bar}$$

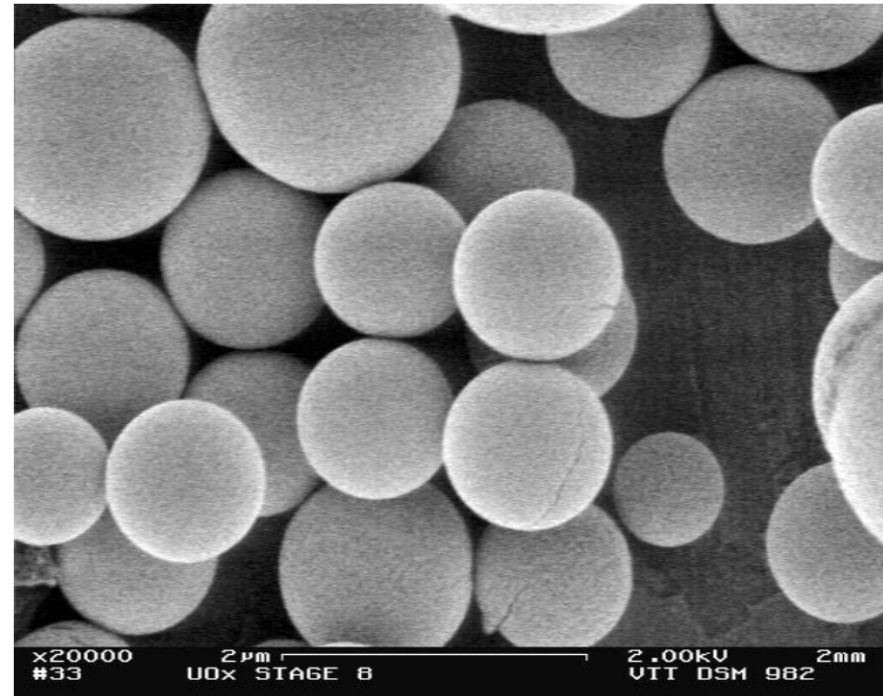
$$p_{H_2O} = 1.5e-7 \text{ bar}$$



© Kleykamp, JNM 131:221, 1985

Example: nuclear fuel aerosol particle preparation

- U and Pu oxide particles for safeguard purposes
- Formation of oxides in the furnace
- Separation by size with Berner Low Pressure Impactor
- Particle average diameter 1 μm



Future plans

- Start UO_2 and simulated nuclear fuel sample annealings in hydrogen atmosphere
- Initiate separate effect tests, first concentrating on a single effect, for example
 - Implications of oxygen partial pressure on the crystal structure and properties
 - Thermal diffusion of elements / particles in the fuel, precipitation
 - Cracking under temperature loads
- Apply microscopy studies along with the separate effect tests
 - Sample preparation for alpha-active materials

VTT

bey⁰nd

the obvious

Janne Heikinheimo
janne.heikinheimo@vtt.fi

@VTTFinland

www.vtt.fi

30/10/2019