

Intercalibration Exercise for Difficult-To-Measure Radionuclides in Activated Steel

Anumaija Leskinen

SYP2019

October 2019

Content

- Context
- Phases of the project
 - 1st phase - Preparation
 - 2nd phase - Radiochemical analysis
 - 3rd phase – Methodology and analysis of the results
- Conclusions

A decorative vertical panel on the left side of the slide, featuring a repeating geometric pattern of interlocking shapes in various shades of blue and green. The pattern includes circles, diamonds, and triangles, creating a complex, tessellated effect.

Context

- Difficult-to-measure (DTM) vs easy-to-measure radionuclides
 - Term used in the decommissioning field
 - What makes them difficult?
 - Formation of scaling factors
- Lack of reference materials
- Validation of radiochemical analysis via intercalibration exercise
- NKS platform
 - Nordic countries have decommissioning projects
 - “DTM Decom” project for DTM analysis in activated steel



Phases of the project

1st phase - preparation

- Participants
 - Finland – VTT (coordinator), Fortum power and heat, Helsinki University
 - Sweden – Cyclife
 - Denmark – Danish technical University
 - Norway – IFE Halden and Kjeller
 - France – CEA Saclay (self funded)
- Online kick-off meeting in February
 - Material to be studied - activated reactor pressure vessel steel
 - Each participant to receive 2-3 samples
 - Main DTMs of interest ^{55}Fe and ^{63}Ni
 - Optional ^{14}C , ^{59}Ni , (^{60}Co)

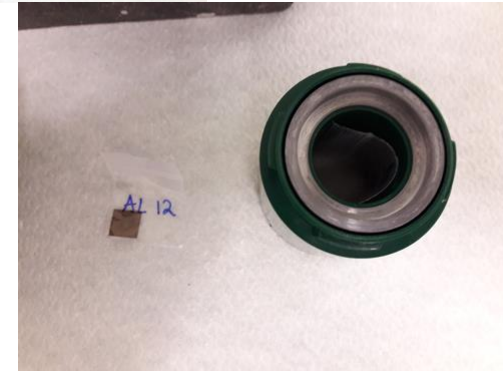
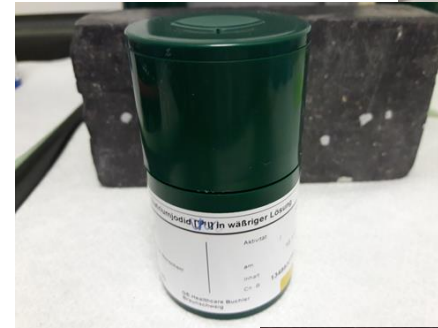
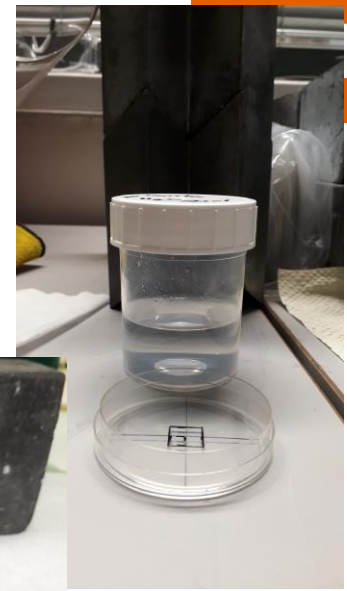
1st phase - preparation

■ Sample preparation

- One RPV steel bar (10 cm x 1 cm x 1 cm) cut to ~0.2 mm slices
- Oxide layer removal, weight after air drying
- Homogeneity studies using gamma spectrometry (ISOCS)
 - Sample holder
 - 3 cm above detector to reduce coincidence
 - 1.8% RSD% of Co-60 in 20 sample measurements
→ homogenous

■ Sending of the samples

- 2-3 samples to each partner
- Sending as UN2910 shipment
- Participant received the packages in May 2019
- Radiochemical analysis time May-September 2019



2nd phase – radiochemical analysis

- Main steps of the radiochemical analysis of ⁵⁵Fe and ⁶³Ni

Step 1 - Decomposition of the solid

Carriers

- Some added carriers
- Some added also hold back carriers

Hot plate

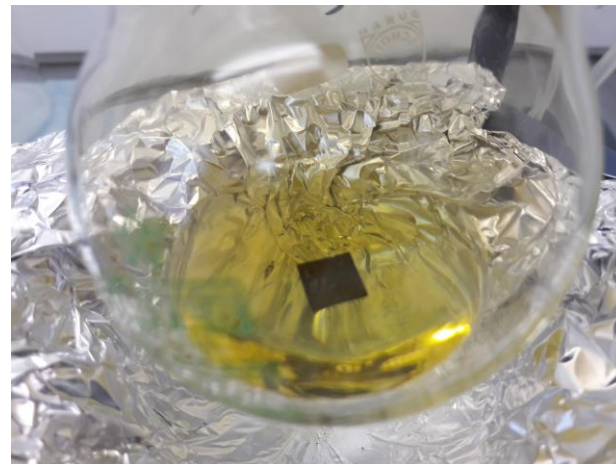
- conc HNO₃:HCl (2:1)
- Aqua regia

Heating mantle + round bottom flask

- when ¹⁴C was analysed
- 5M H₂SO₄ + conc HNO₃
- H₂O:HNO₃:HCl + HClO₄

Microwave

- HNO₃:HF:HCl



2nd phase – radiochemical analysis

- Main steps of the radiochemical analysis of ^{55}Fe and ^{63}Ni

Step 2 - separation of analytes

Solution

- Small fraction (0.3 ml to 1 ml) of the acid digestion solution
- Whole solution

Carriers

- Some added carriers
- Some added hold back carriers

Separation via hydroxide precipitation

- NaOH or NH_4OH to precipitate Ni and Fe (or only Fe)
- Separation of Fe and Ni using anion exchange resin (AG 1x4 resin)
- Separation of Fe and Ni using anion exchange resin (Dowex 1x4) in 9:1 acetone : 6 M HCl mixture (removal of Co)

Removal of silver and separation via TRU resin

- solution to dryness and diluted with HNO_3
- silver precipitated as AgCl with 0,1M HCl
- Fe and Ni separated using TRU resin

2nd phase – radiochemical analysis

- Main steps of the radiochemical analysis of ⁵⁵Fe and ⁶³Ni

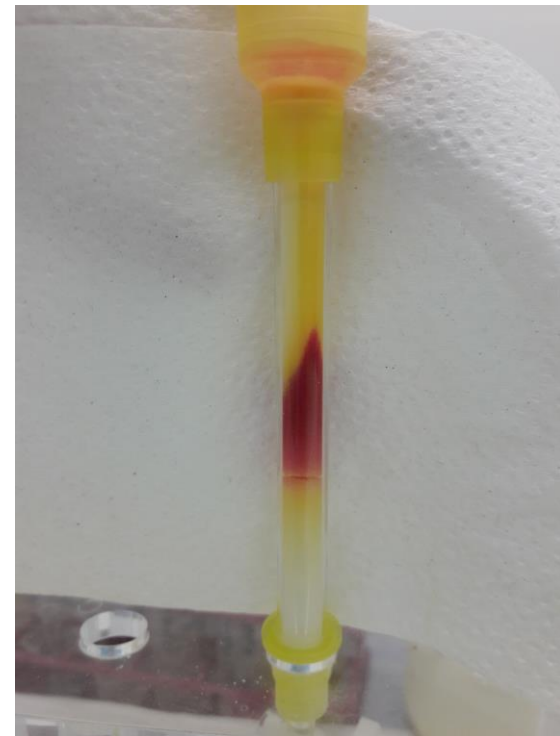
Step 3 - Purification of fractions

Fe fraction

- Some used 2nd AG resin
- Evaporated to dryness
- Dissolved mainly 1M H₃PO₄

Ni fraction

- Purified using Ni-resin (1 or 2 times)
- Evaporated to 0.5 ml – 2 ml



2nd phase – radiochemical analysis

- Main steps of the radiochemical analysis of ⁵⁵Fe and ⁶³Ni

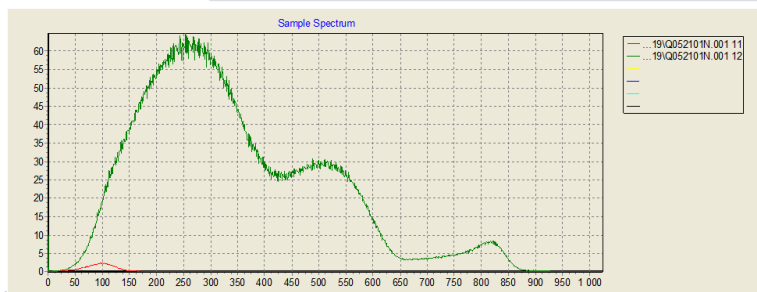
Step 4 - Measurements

LSC fractions

- Measured using LSC
- Standard solutions for quench correction or TDCR technique
- Ultima Gold, HiSafe
- Interference of Co-60 in Ni-63 fraction

Yield fractions

- Measured using ICP-MS, ICP-OES, MP-AES
- Ni yield by spiking a replicate
- Fe yield estimated to be 90%



3rd phase – methodology

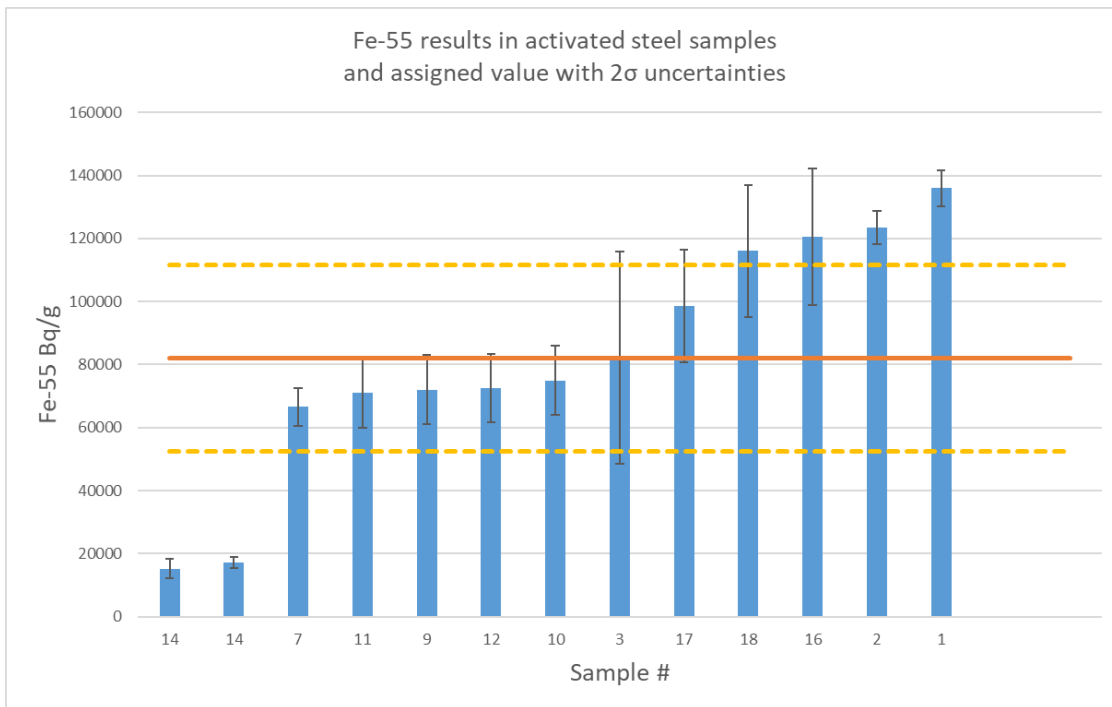
- Analysis based on ISO 13528 standard
- Variety of scoring strategies – most often participant's deviation from an assigned value is compared
- Assigned value x_{pt}
 - Use of real material and not a reference material with assigned values for analytes
 - Assigned value derived from the participant's results
 - Robust mean and standard deviation using Algorithm A
 - Transforms the original data by a process called winsorisation to provide an alternative estimators of mean and standard deviation
 - Expected proportion of outliers is below 20%
- Comparison of performance using z score

Z score	Analysis result
$z \leq 2.0$	Acceptable
$2.0 < z < 3.0$	Warning signal
$z \geq 3.0$	Unacceptable

3rd phase – analysis of the results

■ ⁵⁵Fe results

- 13 entries analysed
- Assigned value: 82.0 ± 29.4 kBq/g (2σ)



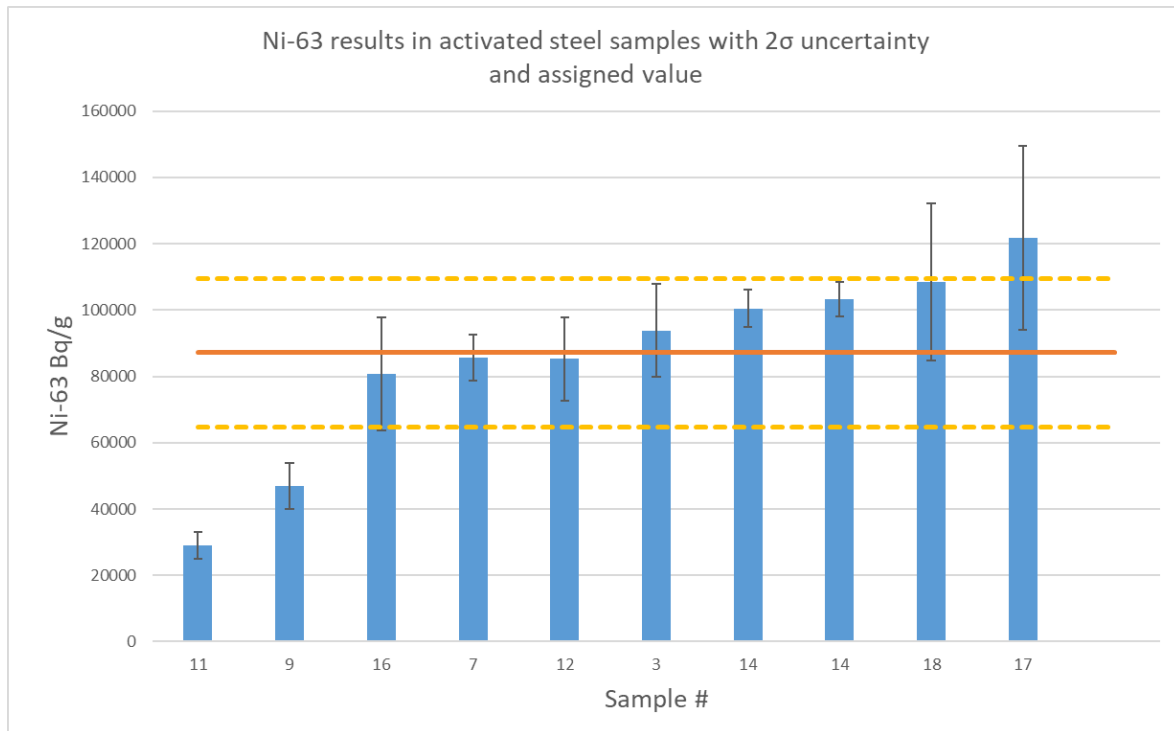
Sample #	Z score
1	3.7
2	2.8
3	0.0
7	1.0
9	0.7
10	0.5
11	0.7
12	0.6
14	4.4
14	4.5
16	2.6
17	1.1
18	2.3

Z score	Analysis result
$z \leq 2.0$	Acceptable
$2.0 < z < 3.0$	Warning signal
$z \geq 3.0$	Unacceptable

3rd phase – analysis of the results

■ ⁶³Ni results

- 10 entries analysed
- Assigned value: 87.1 ± 22.4 kBq/g (2σ)



Sample #	Z score
3	0.6
7	0.1
9	3.6
11	5.2
12	0.2
14	1.5
14	1.2
16	0.6
17	3.1
18	1.9

Z score	Analysis result
$z \leq 2.0$	Acceptable
$2.0 < z < 3.0$	Warning signal
$z \geq 3.0$	Unacceptable

Conclusions

Conclusions

- Preparatory phase needs to be carefully designed
 - DTMs present
 - Sending of the sample
- Radiochemical analyses are similar with small differences
- Assigned value
 - all results were taken into the robust mean analysis
 - possible to make a sub selection of the results – would the results differ?
- Results will be published in NKS website DTM Decom final report and as a peer reviewed publication
- DTM Decom II on activated concrete on 2020 applied

■ Funding

- Nordic Nuclear Safety Research – NKS
- Finnish research program on nuclear waste management - KYT

■ Radioanalytical work was carried out in Centre for Nuclear Safety

■ Radiochemical analysis references

- Gautier, C. et al., (2015) A comparative study using liquid scintillation counting to determine ^{63}Ni in low and intermediate level radioactive waste, *Journal of Radioanalytical and Nuclear Chemistry*, 308:1
- Hazan, I., Korkisch, J. (1965) Anion-exchange separation of iron, cobalt and nickel. *Analytica Chimica Acta*, 32:46-51.
- Hou, X.L. (2005) Determination of ^{14}C and ^3H in Reactor Graphite and Concrete for Decommission, *Appl. Radiat. Isotop.*, 62:871-882
- Hou, X. NKS-B RadWorkshop Risø October 2018, Analytical procedure for simultaneous determination of ^{63}Ni and ^{55}Fe
- Hou, X., et al. (2005) Determination of ^{63}Ni and ^{55}Fe in nuclear waste samples using radiochemical separation and liquid scintillation counting, *Anal. Chim. Acta* 535(1-2): 297-307.
- Hou, X.L., Østergaard L.F., Nielsen S.P. (2005) Determination of ^{63}Ni and ^{55}Fe in nuclear waste and environmental samples, *Anal. Chim. Acta*, 535:297-307
- Hou, X.L., Østergaard L.F., Nielsen S.P. (2007). Determination of ^{36}Cl in Nuclear Waste from Reactor Decommissioning, *Anal. Chem.*, 79:3126-3134
- Leskinen, A., Salminen-Paatero, S., Rätty, A., Tanhua-Tyrkkö, M., Iso-Markku, T., Puukko, E. (accepted October 2019) Determination of ^{14}C , ^{55}Fe , ^{63}Ni and gamma emitters in activated RPV steel samples - a comparison between calculations and experimental analysis, *Journal of Radioanalytical and Nuclear Chemistry*

bey⁰nd

the obvious

Anumaija Leskinen
Anumaija.Leskinen@vtt.fi

www.vtt.fi